## 2025 Fall Semester

# Syllabus for General Chemistry I-Chemical Principles

1. Course: General Chemistry I (CH.10001) [lecture: Experiment: Credit = 3:0:3]

#### 2. Lecture Timetable

Time (Tuesdays and Thursdays)	Class	Professor	Lecture Room(E11)
10:30~12:00 or 14:30~16:00	ТВА	ТВА	ТВА

#### 3. Summary of Lecture

This General Chemistry I lecture offers an overview of fundamental chemistry topics, including a classical approach to chemical bonding, an introduction to quantum mechanics, and a quantum mechanical perspective on atomic and molecular structures. Additional topics are covered to enhance chemical knowledge and deepen understanding of core chemical principles.

#### 4. Material for Teaching:

o Principles of Modern Chemistry, 8th ed, Oxtoby/Gillis/Campion (Brooks/Cole) o Lecture materials will be provided through the KLMS website of each class (https://klms.kaist.ac.kr/).

### 5. General Guidelines

All basic lecture notes can be downloaded at the General Chemistry Website: <a href="http://www.gencheminkaist.pe.kr">http://www.gencheminkaist.pe.kr</a> or a link be found at <a href="http://chem.kaist.ac.kr">http://chem.kaist.ac.kr</a>.

- 1) Practice Sessions led by TAs are scheduled from 8:00 to 8:50 pm on Mondays. These sessions are optional, providing an opportunity for students who seek additional discussion and problem-solving to participate.
- 2) The grading system will be determined based on the total scores achieved by students. The distribution of A grades (including A+, Ao, and A-) will be less than 50% of the total class. A C- grade will correspond to a total score of approximately 50 points. Students taking the course with a P/NR grading option must achieve a score higher than 50 points (equivalent to a C-).

- 3) Grading Criteria and Points Distribution (Total 100 points)
- I. Mid-term Exam: 34 points II. Final Exam: 34 points

(Out of 8-10 questions: 1-2 will come from assignments, 1-2 from previous exams)

III. Homework: 16 points

- Chapter summary: 8 points (1 point for each chapter, maximum of 3 pages, only handwritten assignments will be accepted, each submission is awarded either 0.5 or 1 point, depending on the content)
- Chapter problem: 8 points (1 point for each chapter, each submission is awarded either 0.5 or 1 point, based on the answers provided)

### IV. Attendance & Attitude: 16 points

- Maximum of 16 points (1 point for each attendance of lectures and practice sessions)
- This course does not penalize absences, so there is no recognized attendance.

#### V. Plagiarism of Homework:

- First instance: Warning with a deduction of 10 points, second instance: F grade

#### 6. Waiver Examination

The waiver examination on General Chemistry I will be held at the beginning of the semester, but only for those, who did not take any previous General Chemistry I classes.

## 7. Lecture Schedule

Week (Tuedays, Thursdays)	Chapters	Topics	Due date for Homework (Chap. Summary & problem)	Practice session (Mon, 20:00 ~ 20:50)	Notes
1 <sup>st</sup> (9/2, 9/4)	3	Atomic Shells and Classical Models of Chemical Bonding	-	-	
2 <sup>nd</sup> (9/9, 9/11)	4	Introduction to Quantum Mechanics		0	
3 <sup>rd</sup> (9/16, 9/18)	4,5	Introduction to Quantum Mechanics	Chap3 (Fri, ~23:59)	0	
4 <sup>th</sup> (9/23, 9/25)	5	Quantum Mechanics and Atomic Structure	Chap4 (Fri, ~23:59)	0	
5 <sup>th</sup> (9/30, 10/2)	5, 6	Quantum Mechanics and Atomic Structure	Chap5 (Fri, ~23:59)	0	
6 <sup>th</sup> (10/7, 10/9)					Chuseok and Hangul Proclamation Day
7 <sup>th</sup> (10/14, 10/16)	6	Quantum Mechanics and Molecular Structure	Chap6 (Fri, ~23:59)	0	·
8 <sup>th</sup> (10/21, 10/23)	Mid-term Exam	Chapter 3, 4, 5, 6			
9 <sup>th</sup> (10/28, 10/30)	12	Thermodynamic Processes and Thermochemistry	-		
10 <sup>th</sup> (11/4, 11/6)	12 and 13	Thermodynamic Processes and Thermochemistry / Spontaneous Processes and Thermodynamic Equilibrium		-	
11 <sup>th</sup> (11/11, 11/13)	13	Spontaneous Processes and Thermodynamic Equilibrium	Chap 12(Fri, ~23:59)	0	
12 <sup>th</sup> (11/18, 11/20)	17	Electrochemistry	Chap 13 (Fri, ~23:59)	0	
13 <sup>th</sup> (11/25, 11/27)	17	Electrochemistry	-	0	
14 <sup>th</sup> (12/2, 12/4)	18	Chemical Kinetics	Chap 17 (Fri, ~23:59)	0	
15 <sup>th</sup> (12/9, 12/11)	18	Chemical Kinetics	Chap 18) (Fri, ~23:59)	0	
16 <sup>th</sup> (12/16)	Final Exam				

# 8. Chapter Problems: To be announced